

## NAPHTHALENE DERIVATES FROM BAWANG TIWAI BULB *Eleutherine bulbosa* IN BORNEO

SUPRIANTO SALAM<sup>1,3</sup>, BASO DIDIK HIKMAWAN<sup>1</sup>, ERWIN SAMSUL<sup>1</sup>, HERMAN<sup>2</sup>, ISLAMUDIN AHMAD<sup>2</sup>,  
LAODE RIJAI<sup>2</sup>, AND MOHD AZLAN NAFIRAH<sup>3\*</sup>

<sup>1</sup>Faculty of Pharmacy, Universitas Mulawarman, Kota Samarinda, Kalimantan Timur, Indonesia

<sup>2</sup>Pharmaceuticals Research and Development Laboratory of Pharmaca Tropics, Faculty of Pharmacy,  
Universitas Mulawarman, Kota Samarinda, Kalimantan Timur, Indonesia

<sup>3</sup>Department of Chemistry, Faculty of Science and Mathematics, Sultan Idris Education Universiti, Kota  
Tanjong Malim, Perak, Malaysia

\*Corresponding Author email: azlan@fsmu.upsi.edu.my

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### Abstract

Naphthalene derivative compounds named Eleutherol (**1**) and Eleutherol C (**2**) have been isolated from Bawang Tiwai Bulb *Eleutherine bulbosa* with moderate cytotoxic activity against T47D breast cancer cells. This research was conducted to evaluate cytotoxic activity against T47D breast cancer cells, the compounds Eleutherol (**1**) and Eleutherol C (**2**) are naphthalene derivatives from *Eleutherine bulbosa*, where previously it has been reported that naphthalene derivative compounds have activity against breast cancer cells, especially T47D. After isolating the compounds Eleutherol (**1**) and Eleutherol C (**2**) from the EtOH extract of Bawang Tiwai Bulb *Eleutherine bulbosa* using conventional chromatography methods, their cytotoxic activity was tested against T47D breast cancer cells in vitro. Their chemical structures were elucidated based on spectroscopic analysis including IR, HR-TOFMS, 1D, and 2D NMR, and by comparison to those related spectra previously reported. Two naphthalene derivatives named Eleutherol (**1**) and Eleutherol C (**2**) have been isolated from the Bawang Tiwai Bulb of *Eleutherine bulbosa*. Compounds **1** and **2** were tested for their cytotoxic effects against T47D breast cancer cells and showed moderate cytotoxicity against T47D breast cancer cells with IC<sub>50</sub> values 117.15 and 80.21 μM, respectively, compared with cisplatin 24.07 μM. This research shows moderate cytotoxic activity of the compounds Eleutherol (**1**) and Eleutherol C (**2**) from the EtOH extract of Bawang Tiwai Bulb *Eleutherine bulbosa*. Testing of the cytotoxic effects on T47D breast cancer cells of compounds **1** and **2** of naphthalene derivatives was first reported.

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## INTRODUCTION

*E. bulbosa* is an herbaceous plant belonging to the Iridaceae family that is distributed mainly in South Africa, South America, and Southeast Asia and grows mostly in sulfur areas, 600–2000 m above sea level [1]. In Indonesia, *E. tubers* are widely consumed by local tribal communities in Kalimantan, namely the Dayak tribe, and are traditionally used as a medicinal plant to increase breast milk production and treat diabetes, breast cancer, stroke, hypertension, and sexual disorders [2]. Research has been conducted to identify the compounds responsible for the various properties of *E. bulbosa tubers*. Naphthalene, anthraquinone, and naphthoquinone are the main constituents of *E. bulbosa* [3,4] which exhibits various

pharmacological properties such as anti-diabetic [2], anti-inflammatory [5], anti-microbial [6], anti-melanogenesis [7] and anti-cancer [8]. As an anti-cancer, much research has been carried out on isolated compounds from *E. bulbosa*. For example, the compounds Eleutherinoside C and Isoeleutherin showed selective cytotoxic activity and inhibited TCF/β-catecin transcription in SW480 cancer cells in a dose-dependent manner [8], the compound isoeleutherol showed potent activity in both cancer cell lines against HeLa and MCF-7 cells with LC<sub>50</sub> values of 35.4 and 23.8 ppm respectively [9]. This study is a report on naphthalene derivative compounds that were isolated from the ethanolic extract of *E. bulbosa*. Eleutherol (**1**) and Eleutherol C (**2**) which are derivatives of naphthalene compounds have been

isolated from *E. bulbosa*, along with their cytotoxic activity against T47D breast cancer cells *in vitro*. Activity tests were carried out on T47D because this naphthalene group has very good activity against cytotoxic, especially breast cancer cells.

## EXPERIMENT

### Material and Methode

#### General Experimental Procedures

The UV spectrum was measured using a TECAN Infinite M200 pro (Mannedorf, Switzerland) with MeOH. The IR spectra were recorded on SHIMADZU IR Prestige-21 in KBr (Kyoto, Japan). The high-resolution of time-of-flight mass spectrometry (HR-TOFMS) data was recorded with Water Xevo QTOF MS (Milford, Massachusetts, USA). NMR spectrum was evaluated using JEOL ECZ-500 (Tokyo, Japan) at 500 MHz for  $^1\text{H}$  and 125 MHz for  $^{13}\text{C}$  using tetramethylsilane (TMS) as the internal standard. In addition, the chemical shifts are expressed in ppm, concerning the  $\text{CDCl}_3$  ( $\delta_{\text{H}}$  7.26/ $\delta_{\text{C}}$  77.2) signals. Column chromatography (CC) was performed using silica gel (70-230 mesh) (Merck, Darmstadt, Germany). Thin layer chromatography (TLC) was performed with silica gel 60. TLC plates were precoated with silica gel GF<sub>254</sub> (Merck, 0,25 mm, Darmstadt, Germany) using various solvent systems. Compounds were visualized under UV light (254 and 365 nm) or by spraying the heated silica gel plates with 10%  $\text{H}_2\text{SO}_4$  in EtOH.

#### Plant Material

The bulbs of *E. bulbosa* were collected from Batu Cermin, Samarinda, East Kalimantan, Indonesia. The plant was identified at the Laboratory of Ecology of Tropical Forest Biodiversity, Faculty of Forestry, Mulawarman University, Samarinda, East Kalimantan, Indonesia. The plant was shade-dried (<45°C), coarsely powdered, and stored in an airtight container.

#### Extraction And Isolation

The dried *E. bulbosa* (200 g) was extracted with distilled ethanol at room temperature for 4 days (4×2.5 L) and concentrated using a vacuum rotary evaporator, yielding a concentrated extract (24 g). About 20 g of ethanol was chromatographed using silica gel column CC (70-230 mesh) with *n*-

hexane: ethyl acetate as eluent (5% stepwise) and obtained 5 combined fractions (Fraction A-E).

Fraction A in the isocratic column chromatography and eluted using *n*-hexane: ethyl acetate (9:1) obtained fractions A1-A2. The A2 fraction was purified by recrystallization to obtain compound **1** (7.5 mg).

Fraction B was separated by silica gel column chromatography (70-230 mesh) with a gradient of 2.5% using *n*-hexane: ethyl acetate (9:1-7:3) eluent to produce 5 fractions (Fraction B1-B5). Then fraction B3 was selected to be purified using silica gel column chromatography with *n*-hexane: ethyl acetate (8.5:1.5) as eluent to obtain compound **1** (16.3 mg).

Fraction C was chromatographed on a silica gel column and eluted using *n*-hexane: ethyl acetate with a gradient of 2.5% (9:1-6:4) to produce 9 fractions (Fraction C1-C9). The C4 fraction was selected to be purified using column chromatography. Silica gel was eluted isocratically using *n*-hexane: ethyl acetate (7.5:2.5) and obtained compound **2** (14.2 mg).

*Eleutherol* (**1**) a needle-shaped white crystal (MeOH); IR (KBr)  $V_{\text{max}}$  3471, 1664, 1382, 1320, 1243  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  data see **Table.1**; HR-TOFMS  $[\text{M} + \text{H}]^+$   $m/z$  245.1351 (calcd.  $\text{C}_{14}\text{H}_{13}\text{O}_4$   $m/z$  245.1361).

*Eleutherol C* (**2**) a yellow amorphous powder (MeOH); IR (KBr)  $V_{\text{max}}$  3471, 1664, 1382, 1320, 1243  $\text{cm}^{-1}$ ;  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  data see **Table.1**; HR-TOFMS  $[\text{M} + \text{H}]^+$   $m/z$  245.1351 (calcd.  $\text{C}_{14}\text{H}_{13}\text{O}_4$   $m/z$  245.1361).

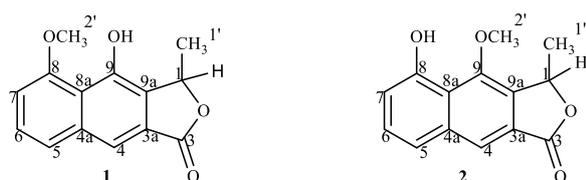
#### Bioassays for Cytotoxic Activity

The T47D cells were seeded into 96-well plates at a density of  $3 \times 10^4$  cells/well and incubated in a humidified  $\text{CO}_2$  incubator for 24 h. Varying concentrations of the test compound were dissolved in DMSO, followed by six desirable concentrations prepared using PBS (phosphoric buffer solution, pH 7.30-7.65). each concentration of the compound was added to the wells in triplicate and incubated in a humidified  $\text{CO}_2$  incubator for 48 h, the negative control wells received only DMSO, and cisplatin was used as the positive control. After an incubation period, MTT reagent [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide] was added to all test and control well and the incubation was continued for another 4 h. The MTT-stop solution containing SDS (sodium dodecyl sulphate) was added to wells and incubated again for another 24 h. The absorbance values were read using a microplate reader at a wavelength of

550 nm. IC<sub>50</sub> values were determined by the linear regression method using Microsoft Excel software. The IC<sub>50</sub> is the concentration required for 50% growth inhibition.

## RESULT AND DISCUSSION

The ethanol extract of *E. bulbosa* was isolated using conventional column chromatography (CC) packed with silica gel 60 with gradient and isocratic elution and spectroscopic analysis to obtain the naphthalene derivatives, Eleutherol (**1**) and Eleutherol C (**2**) (**Figure 1**).



**Figure 1.** Structures of Compounds 1 and 2.

Compound (**1**) was obtained as a needle-shaped white crystal soluble in methanol. The molecular formula was C<sub>14</sub>H<sub>12</sub>O<sub>4</sub> (**Figure 2**), based on the high-resolution time-of-flight (HR-TOFMS) spectrum with a [M + H]<sup>+</sup> ion peak at *m/z* 245.1351 (calcd. C<sub>14</sub>H<sub>13</sub>O<sub>4</sub> *m/z* 245.1361) and obtained Nine degrees of unsaturation. The Infrared (IR) spectrum (**Figure 3**) showed the presence of hydroxyl groups (*V*<sub>max</sub> 3471 cm<sup>-1</sup>), lactone functional groups (*V*<sub>max</sub> 1664 cm<sup>-1</sup>), and Aromatic carbon (*V*<sub>max</sub> 1382, 1320, 1243 cm<sup>-1</sup>). The Proton nuclear magnetic resonance (<sup>1</sup>H-NMR) spectrum (**Figure 4**, **Table 1**) showed one methyl proton signal at δ<sub>H</sub> 1.725 (3H, d, *J* = 6 Hz, CH<sub>3</sub>-1'), one proton methine signal at δ<sub>H</sub> 5.72 (1H, q, *J* = 6.5 Hz, H-1), one hydroxyl chelate signal at δ<sub>H</sub> 9.63 (1H, s, OH-9), one oxygenated methyl proton signal at δ<sub>H</sub> 4.09 (3H, s, OCH<sub>3</sub>-2'), one isolated aromatic proton signal at δ<sub>H</sub> 7.83 (1H, s, H-4) and three aromatic proton signals at δ<sub>H</sub> 6.91 (1H, d, *J* = 7.5 Hz, H-7), δ<sub>H</sub> 7.39 (1H, t, *J* = 8.5 Hz, H-6) δ<sub>H</sub> 7.54 (1H, d, *J* = 8.5 Hz, H-5) which implied the presence of a benzene ring in (**1**). Carbon nuclear magnetic resonance (<sup>13</sup>C-NMR) along with distortions enhancement by polarization transfer (DEPT) (**Figure 5**) showed the presence of fourteen carbon signals consisting of five quaternary carbons *sp*<sup>2</sup> at δ<sub>C</sub> 117.5 (C-8a), 125.9 (C-9a), 127.9 (C-3a) and 137.2 (C-4a). Two quaternary carbons are oxygenated at δ<sub>C</sub> 149.2 (C-9) and 156.6 (C-8). Four carbons of methine at δ<sub>C</sub> 106.3 (C-7), 116.5 (C-4), 123.7 (C-5) and 126.7 (C-6). One carbon of methine is oxygenated at δ<sub>C</sub> 77.5 (C-1). One methyl at δ<sub>C</sub> 19.2 (C-1'), one

oxygenated methyl at δ<sub>C</sub> 56.5 (C-2'), and carbon lactone at δ<sub>C</sub> 170.7 (C-3). These functionalities accounted for six out of the total nine degrees of unsaturation. The remaining three degrees of unsaturation were consistent with a naphthalene skeleton. The IR spectra, as well as the NMR, data indicated (**1**) is a substituted naphthalene derivative [10] which was further confirmed by 2D NMR spectra (**Figure 6**, **Figure 7**). The connectivity of (**1**) was established mainly by proton multiple bond connectivity (HMBC) (**Figure 8**). The HMBC spectrum showed the signal aromatic at δ<sub>H</sub> 7.39 (1H, t, *J* = 8.5 Hz, H-6) was correlated to δ<sub>C</sub> 106.3 (C-7) and 123.7 (C-5) which suggested that the aromatic proton was located at C-6. The methoxyl proton at δ<sub>H</sub> 4.09 (3H, s, OCH<sub>3</sub>-2') was correlated to δ<sub>C</sub> 156.6 (C-8) which indicated that the methoxy group was located at C-8, respectively, the hydroxyl proton at δ<sub>H</sub> 9.63 (1H, s, OH-9) was correlated to δ<sub>C</sub> 149.2 (C-9), 125.9 (C-9a) and 117.5 (C-8a), and the aromatic proton at δ<sub>H</sub> 7.83 (1H, s, H-4) was correlated to δ<sub>C</sub> 127.9 (C-3a) and 137.2 (C-4a), which indicated that the hydroxyl group and aromatic proton were located at C-9 and C-4. The methyl proton at δ<sub>H</sub> 1.725 (3H, d, *J* = 6 Hz, CH<sub>3</sub>-1') was correlated to δ<sub>C</sub> 77.5 (C-1) and 127.9 (C-3a), proving the presence of a methyl group located at C-1. The signal of proton methine at δ<sub>H</sub> 5.72 (1H, q, *J* = 6.5 Hz, H-1) was correlated to the carbonyl lactone at δ<sub>C</sub> 170.7 (C-3), which indicated that the lactone ring was formed between C-3a, C-3, and C-1. Based on the analysis of the spectra IR, 1D, and 2D NMR and compared with the previously reported literature, the structure of (**1**) was similar to the known compound Eleutherol, thus the structure of (**1**) was elucidated as shown and named Eleutherol.

Compound (**2**) was obtained as a yellow amorphous powder soluble in methanol. The molecular formula was C<sub>14</sub>H<sub>12</sub>O<sub>4</sub>, based on the high-resolution time-of-flight (HR-TOFMS) spectrum with a [M + H]<sup>+</sup> ion peak at *m/z* 245.1351 (calcd. C<sub>14</sub>H<sub>13</sub>O<sub>4</sub> *m/z* 245.1361) and obtained Nine degrees of unsaturation. The Infrared (IR) spectrum showed the presence of hydroxyl groups (*V*<sub>max</sub> 3471 cm<sup>-1</sup>), lactone functional groups (*V*<sub>max</sub> 1664 cm<sup>-1</sup>), and Aromatic carbon (*V*<sub>max</sub> 1382, 1320, 1243 cm<sup>-1</sup>). The Proton nuclear magnetic resonance (<sup>1</sup>H-NMR) spectrum showed one methyl proton signal at δ<sub>H</sub> 1.725 (3H, d, *J* = 6 Hz, CH<sub>3</sub>-1'), one proton methine signal at δ<sub>H</sub> 5.72 (1H, q, *J* = 6.5 Hz, H-1), one hydroxyl chelate signal at δ<sub>H</sub> 9.63 (1H, s, OH-9), one oxygenated methyl proton signal at δ<sub>H</sub> 4.09 (3H, s, OCH<sub>3</sub>-2'), one isolated aromatic proton signal at δ<sub>H</sub> 7.83 (1H, s, H-4) and three aromatic

proton signals at  $\delta_H$  6.91 (1H, d,  $J = 7.5$  Hz, H-7),  $\delta_H$  7.39 (1H, t,  $J = 8.5$  Hz, H-6)  $\delta_H$  7.54 (1H, d,  $J = 8.5$  Hz, H-5). The  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  spectra seemed identical to those of (1) (Table 1). However, the HMBC spectrum of (2) (Figure 8) displayed the hydroxyl proton at  $\delta_H$  9.63 (1H, s, OH-9) was correlated to  $\delta_C$  156.6 (C-8), which indicated that the hydroxy group was located at C-

8 (Figure 9). The methoxyl proton at  $\delta_H$  4.09 (3H, s,  $\text{OCH}_3$ -2) was correlated to  $\delta_C$  149.2 (C-9), 125.9 (C-9a) proving the presence of methoxy group located at C-9., respectively. Based on these descriptions and comparing the respective spectroscopic evidence with existing publications, the structure of (3) was elucidated and determined as a known compound as Eleutherol C.

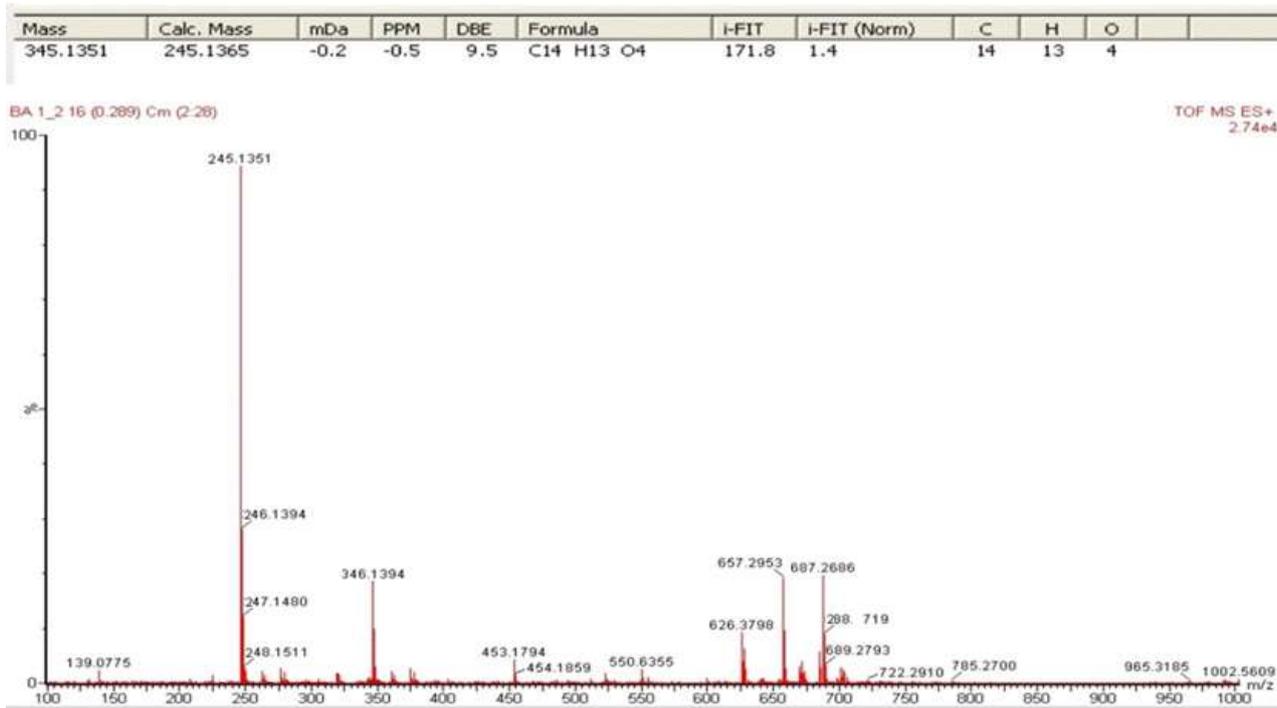


Figure 2. HR-TOFMS spectra of Compound 1.

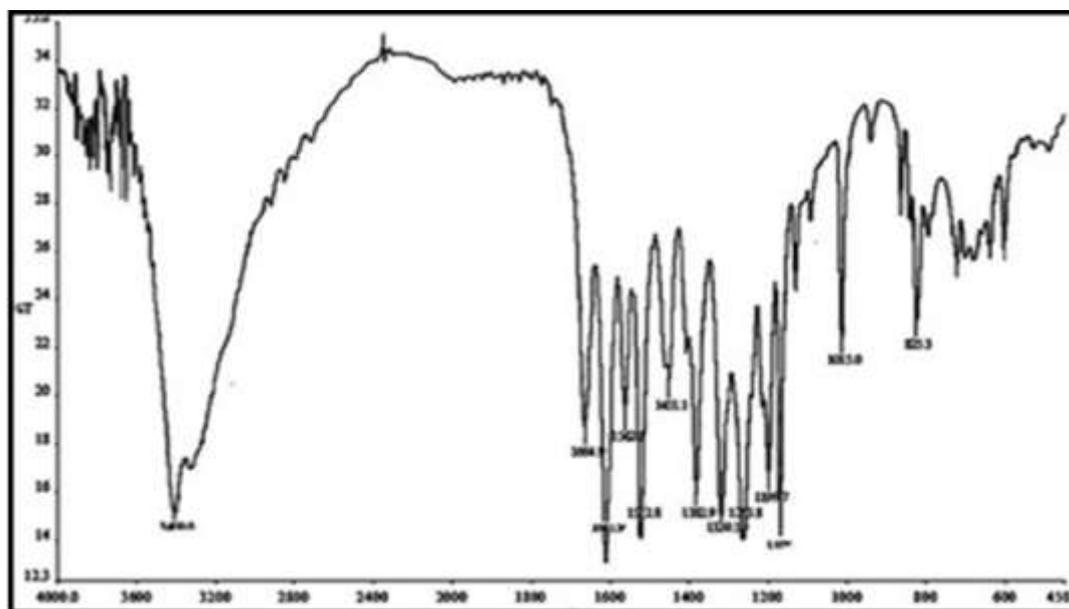


Figure 3. IR spectra of Compound 1 (KBr).

Compounds 1 and 2 were tested for cytotoxic activity against T47D breast cancer cells using the

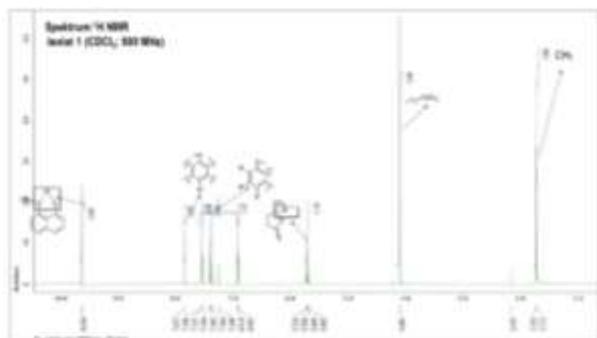
method previously described and cisplatin was used as a positive control. Compounds 1 and 2

showed moderate cytotoxicity with IC<sub>50</sub> values of 117.15 and 80.21 μM, respectively. This indicates that the location of the hydroxyl group can increase cytotoxic activity, whereas in compound **2** the

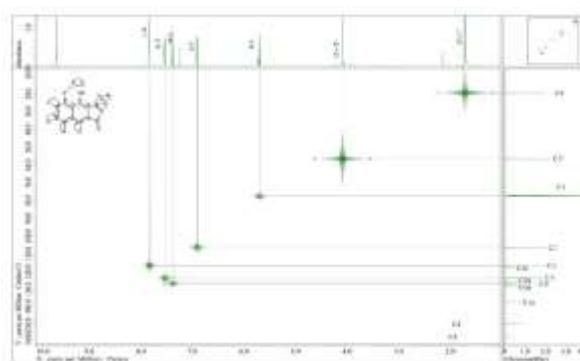
hydroxyl group is located at C-8 and is not blocked by other groups which causes the IC<sub>50</sub> to be lower than compound **1**.

**Table 1.** NMR spectral data for **1** and **2** (500 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C in CDCl<sub>3</sub>).

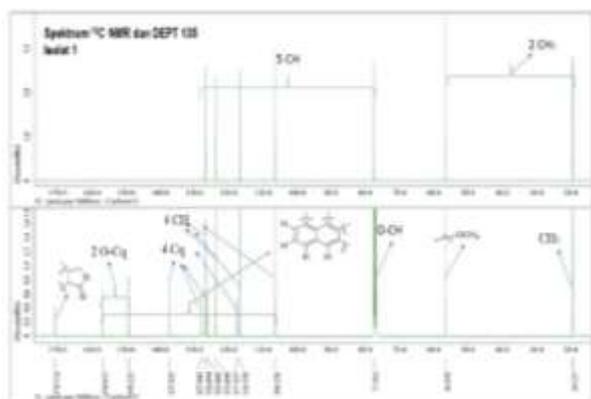
Position Carbon	1		2	
	<sup>13</sup> C-NMR δ <sub>C</sub> (mult,)	<sup>1</sup> H-NMR δ <sub>H</sub> [(ΣH, mult, J(Hz))]	<sup>13</sup> C-NMR δ <sub>C</sub> (mult,)	<sup>1</sup> H-NMR δ <sub>H</sub> [(ΣH, mult, J(Hz))]
1	77.5 (s)	5.72 (1H, q, 6.5)	77.5 (s)	5.72 (1H, q, 6.5)
3	170.7 (q)	-	170.7 (q)	-
3a	127.9 (q)	-	127.9 (q)	-
4	116.5 (s)	7.83 (1H, s)	116.5 (s)	7.83 (1H, s)
4a	137.2 (q)	-	137.2 (q)	-
5	123.7 (s)	7.54 (1H, d, 8.5)	123.7 (s)	7.54 (1H, d, 8.5)
6	126.7 (s)	7.39 (1H, t, 8,5)	126.7 (s)	7.39 (1H, t, 8,5)
7	106.3 (s)	6.91 (1H, d, 7.5)	106.3 (s)	6.91 (1H, d, 7.5)
8	156.6 (q)	-	156.6 (q)	-
8a	117.5 (q)	-	117.5 (q)	-
9	149.2 (q)	-	149.2 (q)	-
9a	125.9 (q)	-	125.9 (q)	-
1'-OCH <sub>3</sub>	19.2 (t)	1.72 (3H, d, 6)	19.2 (t)	1.72 (3H, d, 6)
2'-CH <sub>3</sub>	56.5 (t)	4.09 (3H, s)	56.5 (t)	4.09 (3H, s)
9-OH	-	9.63 (1H, s)	-	-
8-OH	-	-	-	9.63 (1H, s)



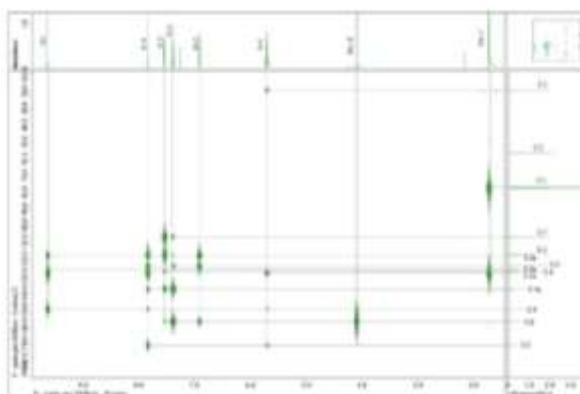
**Figure 4.** <sup>1</sup>H-NMR spectra of Compound **1** (500 MHz, CDCl<sub>3</sub>).



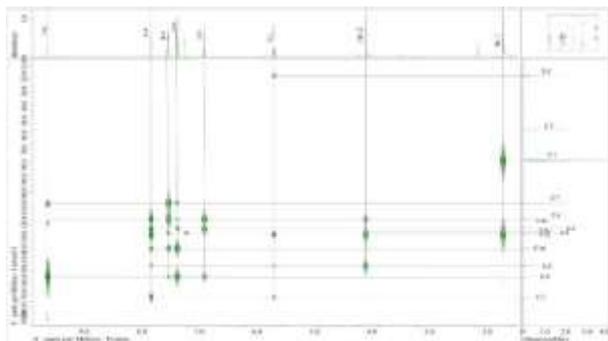
**Figure 6.** HMQC spectra of Compound **1**.



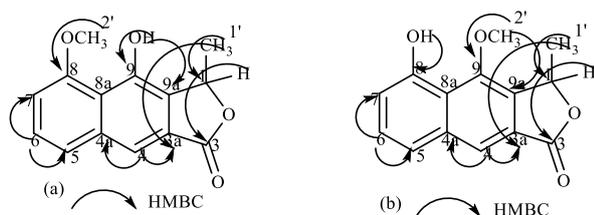
**Figure 5.** <sup>13</sup>C-NMR and DEPT 135° spectra of Compound **1** (125 MHz, CDCl<sub>3</sub>).



**Figure 7.** HMBC spectra of Compound **1**.



**Figure 8.** HMBC spectra of Compound 2.



**Figure 2.** (a) selected HMBC correlation of 1 (b) selected HMBC correlation of 2.

## CONCLUSION

Naphthalene derivatives, namely Eleutherol (1) and Eleutherol C (2) were isolated from the bulbs of *E. bulbosa*. The cytotoxic activity was evaluated against the T47D breast cancer cell line *in vitro*, Compounds 1 and 2 showed weak and no cytotoxic activity with IC<sub>50</sub> values of 117.15 and 80.21  $\mu$ M, respectively compared with cisplatin 24.07  $\mu$ M. Suggesting that the location of the hydroxyl group can increase cytotoxic activity.

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